# **Experimental Investigation of High Strength Geopolymer Concrete under Fire Resistance**

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**Abstract**. Geopolymer concrete is an innovative material that is gaining popularity due to its superior performance and environmental sustainability. The objectives at the current work are to study the effect of slag content, the concentration of alkaline activators on the thermal resistance of geopolymer concrete (GPC). Three mixes of GPC were casted and tested. These mixes are devided into two groups, the first one consists of three mixes: each three slag contents were used 400 kg/m³, tested; for each slag content three different activators percentages were tested; 8%, 10% and 12%. The second group consists of one mix: three different metakaolin replacement ratios were investigated 10%, 15% and 20% for each replacement ratio, two alkaline activators ratios were tested: 10% and 12% while the slag content kept constant for all group mixes at 400 kg/m<sup>3</sup>. For comparison purposes, a control mix of OPC was casted and tested with cement content of 400 kg/m<sup>3</sup>. The whole mixes were exposed to six different temperature 200°C, 300°C, 400°C, 600°C and 800°C for two hours duration. Compressive strength test were carried out after the heat exposure at 28 days age. The results indicated that within the limits of slag content and activator percentages used, the fire resistance increases as the slag content and activator percentages increases. The addition of metakaolin considerable improves the fire resistance of GPC up to 15% replacement ratio. Geopolymer concrete showed superior fire resistance compared to ordinary portland cement (OPC).

Keywords: Geopolymer, Composites, Fire Resistance, Efficiency, Thermal Resistance.

## \. Introduction

Concrete is considered as the most material used in the world during the construction process while the main component used in the concrete is the Ordinary Portland Cement (OPC) as a bonding material. The cement manufacturing releases huge amount of CO<sub>2</sub> in the atmosphere and this leads to the greenhouse gas emission effect [\frac{1}{2}]. Cement production nearly Contributes \frac{1}{2}. \frac{1}{2} of the total CO<sub>2</sub> released in the atmosphere. In general, ordinary Portland cement concrete provides adequate thermal resistance. The ordinary portland cement concrete losses considerable percentage of its strength after exposure to fire [\frac{1}{2}]. This may be attributed to physical and chemical reactions takes place during the temperature rise. As the temperature escalating, the Spalling of Concrete cover takes place and this may lead to direct exposure of steel reinforcement [\frac{1}{2}]. The need of reducing the dependency on OPC as a main construction material due to its big role in global warming and because the huge power consumed in cement industry, encourage the researchers around the world to introduce an alternative for conventional concrete [\frac{1}{2}]. Geopolymer Concrete attracts a great attention in the last two decades [\frac{1}{2}].

One of the main advantageous of geopolymer concrete is the materials used to produce it which are by products and sustainable materials [°]. A wide variety of industrial and agricultural by products can be used as a precursor such as some of ground granulated blast furnace Slag (GGBFS), fly ash, rice husk ash and silica fume ... etc. Natural materials such as metakaolin and bentonite can be used as a precursor to produce geopolymer concrete (GPC) [¹]. Geopolymer is an inorganic aluminosilicate compound produced by the polymerization process takes place as a physico-chemical reaction between the alumino Silicate precursor and alkaline activator such as sodium hydroxide and sodium silicate [¹]. Thermal

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resistance to elevated temperatures is another advantage of geopolymer concrete. Chemical stability of geopolymer concrete made it an excellent alternative for conventional concrete when thermal resistance is a key requirement. Numerous researches were found in the literature concerning the thermal resistance of geopolymer Concrete [A].

Most of the researches concentrate on fly ash and metakaolin based geopolymer. Since these two precursors showed high resistance to elevated temperatures [9]. On the other hand, ground granulated blast furnace slag is the most promising precursor when speaking about compressive strength which is the most important concrete property. In Egypt, there is a resistance of lack of information regarding the thermal resistance of slag based geopolymer concrete produced using our local materials. Based on the above-mentioned discussion, there is great need to fill the gap of information concerning the Subject thermal resistance of slag based geopolymer concrete with local material. In year ' ' ' t, there was IOP Conf. Series [1.]: Materials Science and Engineering 900 to Performance studies on Geo polymer concrete the propose of this is to Strength development is significantly influenced by curing conditions, NaOH molarity, and mix design parameters[11]. Geopolymer concrete shows promise as a sustainable alternative, though its tensile strength needs improvement Although in year 2024 there was Influence of coatings on residual strength of geopolymer concrete columns subjected to fire exposure and the propose is asses the performance of ceramic wool and high alumina cement coating on self-compacting geopolymer concrete (SCGC) columns subjected to ISO 834 fire exposure standards [12] However in year 2024 also International Postgraduate Conference on Materials, Minerals & Polymer And the propose is GPC shows increased compressive strength at 500c due to gepolymerization and the dens matrix of GPC provides superior structural integrity The propose of this study is to make geopolymer concrete with fire resistance with high strength instant of concrete and cement.

## 7. Experimental program

## ۲,۱. Material used

Slag, metakaolin, cement, and sodium hydroxide are considered innovative materials as shown in figs. 1, 7, 7, and 2 respectively that used to produce GPC as shown in Fig. 2.



Fig.\ Slag



Fig. 7 Metakaolin Powder



Fig. Cement Bag



Fig. 5 Sodium Hydroxide



Fig. o GPC Samples

These proposed materials that used to produce different samples of GPC are used in design mix with its allowable properties as shown in Table 1.

Table 1. The properties of the proposed materials.

Materials	Prosperities
Slag	fine grey powder produced by grinding a ground-granulated
3.6 . 1 . 11	blast furnace Indian slag, the specific gravity of slag was 2.9.
Metakaolin	pozzolanic material produced by calcining Kaolinite at high temperature.
Cement	Beni Suef CEM I (52.5) cement was used in control mix, the
	specific gravity of cement equal 3.15.
Sodium Hydroxide	highly soluble in water, the specific gravity of the used Sod. Hydr.is 0.9.
Sodium Silicate	A thick liquid sodium silicate, the specific gravity of Sod. Silicate is 1.2.
sand	Natural clean sand was used as fine aggregate, the fineness modulus of the used sand was 2.89, while its specific gravity was 2.5.
Coarse Aggregates	Clean crushed dolomite with maximum aggregate size of
	20mm, the specific gravity of the used dolomite was 2.6.

## ۲,۲. Superplasticizer

A naphthalene sulphonates - with synthetic polymer- (Adicrete PVF) was used in this research. This kind of superplasticizer allows mixing water condensate de-foamed admixture, the chemical & physical properties of the used superplasticizer are shown in Table <sup>\(\cuparrightar</sup>

Table 7. Chemical and Physical Characteristics of Adecrete BVF.

Cement	SiOr	Al۲O۲	Cao	FerOr	MgO	Na <sub>7</sub> O	KτO	SOr	LOI
Constituents%	۲۰,۱	٥,٣٢	٦١,٧	٣,٦١	۲,۳۹	٠,٣٧	٠,١٣	۲,۸۷	< · , · \

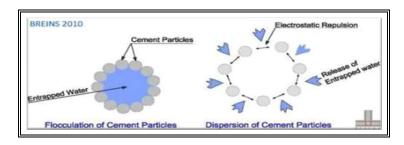


Fig. 7 Schematic representation of a mode of action of naphthalene sulphonate

## ۲, ۳. Samples Preparation

By knowing the specific gravity of each material, the absolute volume equation can used to determine the quantity of aggregate needed in each mix. For example, design of stage \(^1\) mixes will be shown in detail as follows:

$$1000 = \frac{S}{SG_s} + \frac{SP}{SG_{sp}} + \frac{W}{SG_w} + \frac{S.S}{SG_{s.s}} + \frac{S.H}{SG_{S.H}} + \frac{Agg.}{SG_{agg}}$$

Where

S..... Weight of slag in kg used in 'm' of concrete.

SG<sub>s</sub>....Specific gravity of slag.

SP.....Weight of superplasticizer used in 'm' of concrete.

SG.....Specific gravity of superplasticizer.

W.....Weight of water used in 'm' of concrete.

SGw ... Specific gravity of water.

S.S....Weight of sodium silicate used in 'm' of concrete.

SG<sub>S.S</sub>..Specific gravity of sodium silicate.

S.H.....Weight of sodium hydroxide used in 'm' of concrete.

S.G<sub>S.H.</sub>.Specific gravity of sodium hydroxide.

Agg.....weight of aggregate used in 'm' of concrete.

SG<sub>agg</sub>...Specific gravity of aggregate.

## Y, Y, \. General steps to calculate the amount of mix design.

a. Calculate the total amount of Sodium oxide.

$$Na^{\gamma}O = \wedge \%$$
 of total binder content.

$$Na^{\gamma}O = \cdot, \cdot \wedge x \quad \circ \cdot = \gamma \wedge kg.$$

b. Calculate total amount of sodium silicate

Modulus of activator ratio Ms (SiOY / NaYO) = Y

$$\frac{\text{SiO}_2}{\text{Na}_2\text{O}} = 1 \implies \text{SiO}^{\gamma} = \text{Na}^{\gamma}\text{O} = \frac{\xi}{\text{kg}}.$$

$$\text{S.S} = \frac{28}{0.317} = 88.33 \text{ kg}$$

c.Calculate amount of NaYO in sodium silicate.

NaYO in S.S = 
$$\lambda\lambda$$
,  $\gamma\gamma x \cdot , \gamma\gamma = \gamma \cdot , \circ kg$ .

d.Calculate Na O needed.

NaYO needed = 
$$\forall \land - \lor , \circ = \lor \lor , \circ kg$$
.

e. Calculate the total amount of NaOH

$$NaOH = \frac{17.5}{0.6025} = 49.45 \text{ kg}$$

f. Extra water calculations

Water in S.S = 
$$\lambda\lambda$$
,  $\forall x \cdot$ ,  $\circ 170 = \xi 9$ ,  $11 \text{ kg}$ .

Water in NaOH = 
$$\Upsilon9, \cdot \xi \times \cdot, \Upsilon9V0 = 11,0 \text{ kg}$$

Total amount of water = w/b x total binder content =  $\cdot$ ,  $\xi$  x  $ro \cdot = 1 \xi \cdot \text{ liters}$ 

Extra water needed = Total amount of water – water in S.S – water in NaOH

$$= 15. - (79.7 + 11.0) = 94.9$$
 liters.

Using the absolute volume equation, the total amount of agg. is calculated as follows:

$$1000 = \frac{350}{2.8} + \frac{0.4 \times 350}{1.18} + \frac{98.9}{1} + \frac{88.33}{1.2} + \frac{29.04}{0.8} + \frac{Agg.}{2.6}$$

Sand = 
$$\circ$$
97 kg.

Three slag content ( $S^{\xi \cdot \cdot \cdot}$ ) were used and tested in this experimental program as Shown in Table 3.

Table \(^{\text{N}}\). Simple of golypolymer concrete mix design.

Mixes	Slag (Kg)	S.P. (Kg)	Water (Kg)	Sand (Kg)	C. Agg. (Kg)	S.S (Kg)	NaOH (Kg)	M.K
S : A	٤٠٠	١٦	9.,17	011,00	1.71,20	1 , 9	٣٣,٠٣	
S :	٤٠٠	١٦	۲۲,٦١	001,9	١٠٣٨	177,7	٤١,٢٦	
S:17	٤٠٠	١٦	00,10	०१२,१२	1.12,0	101,27	٤٩,٥	

These three mixes in order to investigate the effect of adding metakaolin as a slag supplement, a group with slag content  $\xi \cdot \cdot \text{kg/m}^{\text{m}}$  was chosen. Three different percentages of metakaolin as supplemented, i.e.,  $1 \cdot \text{kg/m}^{\text{m}}$  was chosen. Three different percentages of metakaolin as supplemented, i.e.,  $1 \cdot \text{kg/m}^{\text{m}}$  was chosen. Three different percentages of metakaolin as supplemented, i.e.,  $1 \cdot \text{kg/m}^{\text{m}}$  was chosen. Three different percentages of metakaolin as supplemented, i.e.,  $1 \cdot \text{kg/m}^{\text{m}}$  was chosen. Three different percentages of metakaolin as supplemented, i.e.,  $1 \cdot \text{kg/m}^{\text{m}}$  was chosen. Three different percentages of metakaolin as supplemented, i.e.,  $1 \cdot \text{kg/m}^{\text{m}}$  was chosen. Three different percentages of metakaolin as supplemented, i.e.,  $1 \cdot \text{kg/m}^{\text{m}}$  was chosen. Three different percentages of metakaolin as supplemented, i.e.,  $1 \cdot \text{kg/m}^{\text{m}}$  was chosen. Three different percentages of metakaolin as supplemented, i.e.,  $1 \cdot \text{kg/m}^{\text{m}}$  was chosen. Three different percentages of metakaolin as supplemented, i.e.,  $1 \cdot \text{kg/m}^{\text{m}}$  was chosen. Three different percentages of metakaolin as supplemented, i.e.,  $1 \cdot \text{kg/m}^{\text{m}}$  was chosen. Three different percentages of metakaolin as supplemented, i.e.,  $1 \cdot \text{kg/m}^{\text{m}}$  was chosen. Three different percentages of metakaolin as supplemented, i.e.,  $1 \cdot \text{kg/m}^{\text{m}}$  was chosen. Three different percentages of metakaolin as supplemented, i.e.,  $1 \cdot \text{kg/m}^{\text{m}}$  was chosen. Three different percentages of metakaolin as a slag supplemented, i.e.,  $1 \cdot \text{kg/m}^{\text{m}}$  was chosen. Three different percentages of metakaolin as a slag supplemented, i.e.,  $1 \cdot \text{kg/m}^{\text{m}}$  was chosen.

#### ۲, ٤. Mixing Procedures

Sand, Coarse aggregate and Slag (or Slag + Metakaolin) were mixed dry first in the mixer for <sup>\tau</sup> minutes. The alkaline solution was prepared by adding sodium hydroxide to sodium silicate and mixing together for <sup>\tau</sup> min., then the water was added. Finally, the superplasticizer was added to the solution. The alkaline solution then added gradually during mixing. The mixing process takes about another <sup>\tau</sup> minutes to reach homogeneous mix.

## Y,o. Casting Procedure

eighteen moulds  $(x) \cdot x \cdot x$  cm were used to cast eighteen concrete cube samples for each mix. After casting the concrete into the moulds, the moulds were comparted using the vibrating table as shown in Fig. (x).



Fig. <sup>\( \)</sup> The vibrating table

### ۲,٦ Curing

All specimens were cured in room temperature until the date of testing.

## Y, V. Thermal Exposure Regime

In order to measure the resistance of geopolymer concrete to thermal exposure,  $^{\tau}$  concrete cubes were tested in room temperature (control or reference samples), for each mix  $^{\tau}$  concrete cubes, for each temperature, were exposed to the following five different temperatures:  $^{\tau} \cdot ^{\circ}C$ ,  $^{\tau} \cdot ^{\circ}C$ ,  $^{\tau} \cdot ^{\circ}C$ ,  $^{\tau} \cdot ^{\circ}C$ , and  $^{\wedge} \cdot ^{\circ}C$ . The mix design of normal concrete is showed in Table  $^{\xi}$ .

Table <sup>₹</sup>. The mix design of normal concrete.

Material	Cement	Sand	Dol size \ \foats \ Dol size		Total Admixture		Fiber
			mm	۲°mm	water	( <b>HG</b> <sup>1</sup> )	glass
Weight	٤٠٠	<b>٦</b> ٨٠	٥٦.	٥٨,	19.	٦	٠,٩
٠,٠٤	17,	77,7	۲۲,٤٠٠	77,7	٧,٦٠٠	٠,٢٤٠	٠,٠٣٦

#### ۳. Results and discussion:

## **7,1.** Effect of temperature on OPC:

The results show that the original compressive strength ( $\circ \cdot \cdot \text{kg/cm}_2$ ) decreases gradually as the heat temperature increases. Increasing the heat temperature to  $? \cdot \cdot \circ C$  leads to  $? \cdot \cdot \circ C$  decrease in compressive strength. Exposing the OPC to  $? \cdot \cdot \circ C$ ,  $? \cdot \cdot \circ C$  and  $? \cdot \cdot \circ C$  yields to decrease in compressive strength by  $? \cdot \cdot \cdot \circ C$  and  $? \cdot \cdot \circ C$  the concrete compressive strength significantly decreased to nearly one third of the original compressive strength as shown in fig.  $? \cdot \cdot \circ C$ 

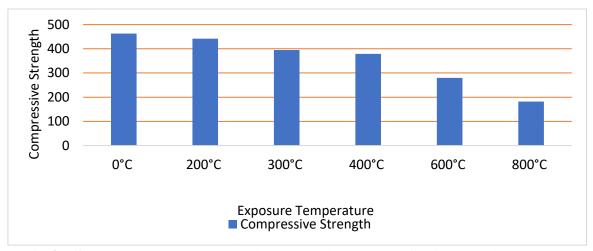


Fig. A Effect exposure temperature on the compressive strength of ordinary portland concrete

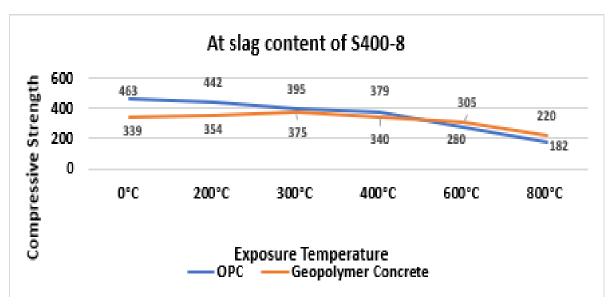


Fig. 9 Effect exposure temperature on the compressive strength of ordinary portland concrete

The influence of exposure temperature on the compressive strength of both OPC and geopolymer concrete with  $1\cdot\%$  activator. The same trend as above was observed. The compressive strength of OPC decreases as the exposure temperature increases by  $\frac{1}{2}\%$ ,  $\frac{1}{2}\%$ ,  $\frac{1}{2}\%$ ,  $\frac{1}{2}\%$ , and  $\frac{1}{2}\%$  as the exposure temperature increases to  $\frac{1}{2}\%$ ,  $\frac{1}{2}\%$ ,  $\frac{1}{2}\%$ ,  $\frac{1}{2}\%$ ,  $\frac{1}{2}\%$ , and  $\frac{1}{2}\%$  as the exposure temperature increases to  $\frac{1}{2}\%$ ,  $\frac{1}{2}$ 

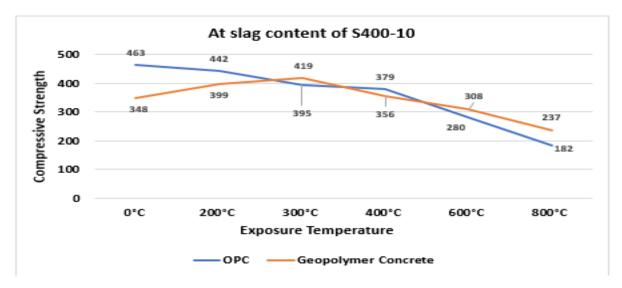


Fig. \( \cdot \) Effect exposure temperature on the compressive strength of ordinary portland concrete

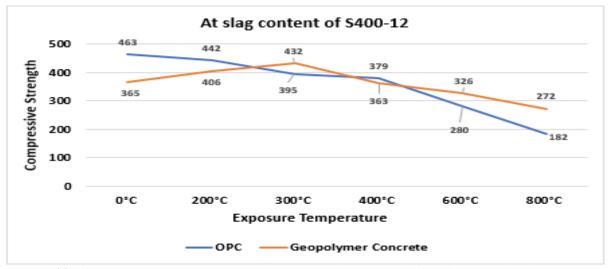


Fig. \( \) Effect exposure temperature on the compressive strength of ordinary portland concrete

#### ٤. Conclusion

It is possible to draw the conclusion that geopolymer concrete exhibits better heat resistance than regular Portland concrete based on the facts and discussion above:

- \. Increasing the slag content significantly improves the heat resistance of geopolymer concrete within the range of slag content studied in this study.
- \*. The compressive strength of geopolymer concrete is significantly increased by thermal **exposure** up to "...°C. This rise could be as much as "". This rise can be the result of improving GPC's polymerization process.
- **▼.** The compressive strength of GPC gradually decreases at temperatures beyond  $\xi \cdot \cdot \cdot \circ C$ .
- £. GPC's compressive strength is greatly impacted by thermal exposure to ^..oC. The remaining strength could be as much as o..o of its initial value. Deboned alumino-silicate and calcium alumino-silicate hydrate may be the cause of this.o. Within the range of activator percentages tested in this thesis, increasing the activator percentage increases the compressive strength of the GPC, and consequently enhances the thermal resistance of GPC. Increasing the activator percentage positively leads to increase the calcium alumino silicate hydrates bond tees.
- 1. The significance of increasing the activator percentage is highly observative for the mixes with %0 kg/m³ and 5 · · kg/m³ slag content. For rich slag mixes with 50 · kg/m³, the significance of increasing the activator percentage decreases to enhance the thermal resistance of GPC.
- <sup>V</sup>. Geopolymer Concrete has excellent thermal resistance properties compared with conventional ordinary portland cement concrete.

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